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# Stereoselective addition of allylstibonium bromide to aldehydes \*

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#### Abstract

Reaction of allylantimony with aldehydes provides homoallylic alcohols with high *threo* selectivity in the case of (E)-4-methyl-2-pentenylantimony (2c) and with preferential *erythro* selectivity in the case of crotylantimony (2a).

## Introduction

Stereoselective synthesis of  $\alpha$ -methylhomoallylic alcohols, of possible application to the synthesis of macrolide and polyether antibiotics and of some pheromones, is one of the most challenging problems for the synthetic chemist [1]. One successful strategy for this purpose involves stereoselective reaction of crotylmetals with aldehydes [2]. Of particular interest is the dependence of the stereoselectivity of crotylstannanes upon the reaction conditions [3]. However, the analogous reaction of allylantimony has hardly been studied [4]. Here we report a diastereoselective addition of allylantimony to aldehydes.

#### **Results and discussion**

Allylantimony 2 was readily obtained by mixing tributylstibine with bromides 1 at room temperature. Heating the salt 2 with a variety of aldehydes under nitrogen produced homoallylic alcohols 3 in high yield. This reaction was performed without any solvent. The reaction can also take place in 1,4-dioxane under reflux with moderate yield and similar diastereoselectivity. However, the reaction was slow in THF under reflux, because of the low boiling point of THF. The ratio of *threo* and *erythro* was determined by <sup>1</sup>H NMR and/or capillary GC analysis by comparison with authentic samples. The results are summarized in Table 1.

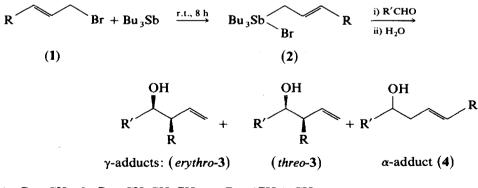
<sup>\*</sup> This paper is the XCIII report on the studies of the application of elemento-organic compounds of the 15th and 16th groups in organic synthesis.

Allylantimony ( <i>trans</i> and/or <i>cis</i> ) <sup>b</sup>	Entry	Aldehyde	Total yields <sup>c</sup> (%)	Products ratio <sup>d</sup>	
				threo/erythro	3/4
$Bu_{3}Sb < Br \\ CH_{2}CH = CHCH_{3}$	a	C <sub>6</sub> H <sub>5</sub> CHO <sup>e</sup>	97	33:67	97:3
	b	p-BrC <sub>6</sub> H <sub>4</sub> CHO	93	35:65	98:2
	с	p-MeC <sub>6</sub> H <sub>4</sub> CHO	92	35:65	98:2
( t 86%, c 14%)	d	p-CIC6H4CHO	92	35:65	98:2
	e	<sup>i</sup> PrCHO	<b>9</b> 0	36:64	78:22
	f	CH <sub>3</sub> CH=CHCHO	92	37:63	94:6
	g	n-C <sub>8</sub> H <sub>17</sub> CHO	95	35:65	<b>95</b> : 5
Br	h	p-ClC <sub>6</sub> H₄CHO	95	44:56	93:7
$Bu_{3}Sb \begin{pmatrix} Br \\ CH_{2}CH = CHC_{3}H_{7} \end{pmatrix}$ (1)	i	p-MeC <sub>6</sub> H₄CHO	95	48:52	91:9
	j	<sup>i</sup> PrCHO	91	65:35	80:20
	k	СН₃СН=СНСНО	93	47:53	94:6
Br	}	p-ClC <sub>6</sub> H₄CHO	94	95:5	85 : 15
$Bu_3Sb < Br CH_2CH = CHCH(CH_3)_2$	m	p-MeC, H <sub>4</sub> CHO	95	95:5	82:18
	n	PrCHO	95	99:1	70:30
(1)	о	CH3CH=CHCHO	93	85:15	90:10

Synthesis of homoallylic alcohols via allylantimony<sup>a</sup>

<sup>*a*</sup> All reactions were performed as described in the text. <sup>*b*</sup> Determined by <sup>1</sup>H NMR and <sup>13</sup>C NMR. <sup>*c*</sup> Isolated yields. <sup>*d*</sup> Determined by <sup>1</sup>H NMR and/or capillary GC analysis by comparison with authentic materials. <sup>*e*</sup> If the reaction was carried out in refluxing 1,4-dioxane, the product was obtained in 60% yield with similar diastereoselectivity. <sup>*f*</sup> The same result was given by tributylstibine, crotylbromide and aldehyde in one pot.

The reaction of 2 with aldehydes results predominantly in  $\gamma$ -adduct as in the case of other crotylmetallic reagents, such as crotyltin [3]. Beside the  $\gamma$ -adduct, some  $\alpha$ -adduct 4 is formed as a by-product. The production of  $\alpha$ -adduct may be attributed to the more enhanced ionic nature of the allylic C-Sb bond, which is different from what applies in crotyltin [5\*].



(a:  $\mathbf{R} = \mathbf{CH}_3$ ; b:  $\mathbf{R} = \mathbf{CH}_3\mathbf{CH}_2\mathbf{CH}_2$ ; c:  $\mathbf{R} = (\mathbf{CH}_3)_2\mathbf{CH}$ )

Table 1

<sup>\*</sup> Reference number with an asterisk indicates a note in the list of references.

The diastereoselectivity of this reaction depends greatly upon the substituent R. In the case of 2a (R = CH<sub>3</sub>), a mixture of *erythro* and *threo* isomers was isolated in a ratio of about 2:1 (entries a-g). In the case of the highly hindered 2c (R = (CH<sub>3</sub>)<sub>2</sub>CH), the *threo* isomer was obtained with 70-98% diastereoselectivity (entries (1-o).

As a result of the enhanced ionic nature of the Br-Sb bond, either the cyclic or acyclic transition state in this reaction could be favoured. In the case of 2a, stibonium bromide may act as a lewis acid, and the propensity to an acyclic transition state seems to be greater than that to a cyclic transition state, consequently erythroselectivity was observed. Otherwise, in the case of 2c, because of the steric properties of iso-propyl the propensity to a cyclic transition state was greater, so a high degree of *threo*-selectivity was observed as in the case of crotyltins [2]. As for 2b, the result was intermediate. However, this mechanistic rationale is speculative. Confirmation of the mechanism of this reaction awaits more detailed understanding of the reaction course.

#### Experimental

IR spectra were obtained on a Schimadzu IR-440 spectrophotometer and are reported in cm<sup>-1</sup> units (neat). Mass spectra were measured on a Finnigan GC-MC 4021 spectrometer. <sup>1</sup>H NMR spectra were recorded on a Varian EM-360 or AM-500 spectrometer in CCl<sub>4</sub> solution unless noted otherwise, with TMS as an internal standard and are reported in  $\delta$  units (ppm).

## 2-Methyl-1-phenyl-3-buten-1-ol (3a entry a) [3b]

Typical procedure: Tributylstibine (675 mg, 2.3 mmol) and crotyl bromide (350 mg, 2.6 mmol) were mixed and stirred at ambient temperature for 8 h under nitrogen. The resulting oily product was heated with benzaldehyde (210 mg, 2.0 mmol) at 100 °C for 15–18 h. After protonolysis with wet alcohol, the mixture was chromatographed on an alumina-silica gel (1:1) column, eluting with 95:5 petro-leum ether/ethyl acetate to give a mixture of  $\alpha$ - and  $\gamma$ -adduct products (310 mg, 97%), b.p. 93–95 °C/1 mmHg. *threo*-Form. <sup>1</sup>H NMR: 0.88 (d,  $J_1 = 7.0$  Hz, 3H); 1.70 (brs, 1H); 2.42 (m, 1H); 4.31 (d,  $J_2 = 7.0$  Hz, 1H); 4.70–5.26 (m, 2H); 5.30–6.10 (m, 1H); 7.26 (s, 5H). IR: 3400, 1640, 1270, 1020, 980, 910, 760, 700 cm<sup>-1</sup>. MS: 162 ( $M^+$ , 0.1), 145 (26), 108 (100), 107 (28), 105 (23), 80 (42), 79 (35), 77 (53). *erythro*-Form. <sup>1</sup>H NMR: 0.95 (d,  $J_1 = 7.0$  Hz, 3H); 4.50 (d,  $J_3 = 6$  Hz, 1H). The other data of <sup>1</sup>H NMR, IR and MS are the same as above.

#### 1-(4-Bromophenyl)-2-methyl-3-buten-1-ol (3a entry b)

From 4-bromobenzaldehyde: 370 mg. Mixture products: 450 mg, 93%. B.p.  $138-140 \degree \text{C}/1 \text{ mmHg. threo-Form.}^{1}\text{H} \text{ NMR: } 0.83 (d, J_1 = 7.0 \text{ Hz}, 3\text{H}); 2.20 (brs, 1\text{H}); 2.35 (m, 1\text{H}); 4.20 (d, J_2 = 7.0 \text{ Hz}, 1\text{H}); 4.70-5.20 (m, 2\text{H}); 5.60 (m, 1\text{H}); 7.06 (d, J_3 = 10.0 \text{ Hz}, 2\text{H}); 7.39 (d, J_3 = 10.0 \text{ Hz}, 2\text{H}). \text{IR: } 3400, 1640, 1010, 920 \text{ cm}^{-1}. \text{MS: } 242, 240 (M^+, 0.1), 225, 223 (14), 187 (84), 157 (19), 77 (100). erythro-Form. ^{1}\text{H} \text{NMR: } 0.92 (d, J_1 = 7.0 \text{ Hz}, 3\text{H}); 4.37 (d, J_4 = 5.5 \text{ Hz}, 1\text{H}). \text{ The other data of } ^{1}\text{H} \text{NMR}, \text{ IR and MS are the same as above. Anal. Found: C, 54.57; H, 5.57. } C_{11}H_{13}\text{BrO}$  (mixture products) calcd.: C, 54.79; H, 5.43%.

## 2-Methyl-1-p-tolyl-3-buten-1-ol (3a entry c) [3b]

From *p*-tolualdehyde: 240 mg. Mixture products: 325 mg, 92%. B.p. 114–117°C/1 mmHg. *threo*-form. <sup>1</sup>H NMR: 0.85 (d,  $J_1 = 6.5$  Hz, 3H); 1.25–2.25 (m, 2H); 2.32 (s, 3H); 4.22 (d,  $J_2 = 7.0$  Hz, 1H); 4.95–5.15 (m, 2H); 5.48–5.80 (m, 1H); 7.08 (bs, 4H). IR: 3610, 1620, 1050, 990, 910 cm<sup>-1</sup>. MS: 176 ( $M^+$ , 0.2), 175 (0.4), 159 (14), 122 (100), 93 (59), 91 (40), 77 (30). *erythro*-Form. <sup>1</sup>H NMR: 0.95 (d,  $J_1 = 6.5$  Hz, 3H); 4.40 (d,  $J_3 = 6.0$  Hz, 1H). The other data of <sup>1</sup>H NMR, IR and MS are the same as above.

#### 1-(4-Chlorophenyl)-2-methyl-3-buten-1-ol (3a entry d) [3a]

From 4-chlorobenzaldehyde: 281 mg. Mixture products: 360 mg, 92%. *threo*-Form. <sup>1</sup>H NMR (CDCl<sub>3</sub>): 0.87 (d,  $J_1 = 7.0$  Hz, 3H); 2.30 (m, 1H); 2.35 (brs, 1H); 4.35 (d,  $J_2 = 6.8$  Hz, 1H); 4.80–5.25 (m, 2H); 5.40–6.10 (m, 1H); 7.20 (s, 4H). IR: 3400, 1640, 1095, 1010, 990, 920 cm<sup>-1</sup>. MS: 196 ( $M^+$ , 0.2), 181 (6), 179 (18), 143 (41), 142 (55), 141 (100), 113 (27), 77 (88). *erythro*-Form. <sup>1</sup>H NMR: 0.98 (d,  $J_1 = 7.0$ Hz, 3H); 4.60 (d,  $J_3 = 5.6$  Hz, 1H). The other data of <sup>1</sup>H NMR, IR and MS are the same as above.

## 2,4-Dimethyl-5-hexen-3-ol (3a entry e) [3b]

From isobutyraldehyde: 144 mg. Mixture products: 230 mg, 90%. threo-Form. <sup>1</sup>H NMR: 0.8–1.1 (m, 9H); 1.4 (m, 1H); 1.64 (brs, 1H); 2.06 (m, 1H); 3.00 (dd,  $J_1 = 5.0, J_2 = 10.5$  Hz, 1H); 4.75–5.16 (m, 2H); 5.68 (m, 1H). IR: 3400, 1630, 1000, 910 cm<sup>-1</sup>. MS: 128 ( $M^+$ , 0.3), 111 (9), 73 (57), 56 (100). erythro-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above.

#### 3-Methyl-1,5-heptadien-4-ol (**3a** entry f) [3b]

From crotonaldehyde: 140 mg. Mixture products: 230 mg, 92%. *threo*-Form. <sup>1</sup>H NMR: 0.97 (d,  $J_1 = 6.6$  Hz, 3H); 1.70 (d,  $J_2 = 5.0$  Hz, 3H); 2.25 (m, 1H); 2.7 (bs, 1H); 3.8 (dd,  $J_3 = 5.5$ ,  $J_4 = 10.0$  Hz, 1H); 4.85–5.05 (m, 2H); 5.40–5.90 (m, 3H). IR: 3600, 1640, 990, 960, 910 cm<sup>-1</sup>. MS: 126 ( $M^+$ , 0.2), 125 (1), 109 (46), 72 (100), 69 (23), 43 (58). *erythro*-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above.

## 3-Methyl-1-dodecen-4-ol (3a entry g) [6]

From nonyl aldehyde: 284 mg. Mixture products: 375 mg, 95%. *threo*-Form. <sup>1</sup>H NMR: 0.85 (m, 6H); 1.15 (brs, 12H); 1.33 (brs, 1H); 1.53 (m, 2H); 2.0 (m, 1H); 3.20 (m, 1H); 4.65–5.05 (m, 2H); 5.55 (m, 1H). IR: 3400, 1640, 990, 910 cm<sup>-1</sup>. MS: 198  $(M^+, 0.4)$ , 181 (0.5), 143 (11), 141 (14), 83 (48), 71 (23), 69 (79), 56 (100). *erythro*-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above.

## 1-(4-Chlorophenyl)-2-propyl-3-buten-1-ol (3b entry h)

From 4-chlorobenzaldehyde: 281 mg. Mixture products: 425 mg, 95%. B.p.  $140-145 \,^{\circ}$  C/1 mmHg. *threo*-Form. <sup>1</sup>H NMR (CDCl<sub>3</sub>): 0.78 (t,  $J_1 = 7.0$  Hz, 3H); 1.13-1.48 (m, 4H); 2.10 (brs, 1H); 2.23 (m, 1H); 4.40 (d,  $J_2 = 6.8$  Hz, 1H); 5.15-5.26 (m, 2H); 5.58-6.68 (m, 1H); 7.25 (m, 4H). IR: 3400, 1640, 1090, 1020, 920, 830 cm<sup>-1</sup>. MS: 224 ( $M^+$ , 0.14), 207 (4), 143 (53), 141 (100), 113 (12), 77 (51). *erythro*-Form. <sup>1</sup>H NMR: 0.86 (t,  $J_1 = 7.0$  Hz, 3H); 2.38 (m, 1H); 4.60 (d,  $J_3 = 5.8$  Hz, 1H); 4.98-5.08 (m, 2H); 5.43-5.50 (m, 1H). The other data of <sup>1</sup>H NMR, IR and MS are the same as above.

## 2-Propyl-1-p-tolyl-3-buten-1-ol (3b entry i)

From *p*-tolualdehyde: 240 mg. Mixture products: 385 mg, 95%. B.p. 130–133° C/1 mmHg. *threo*-Form. <sup>1</sup>H NMR: 0.9 (t,  $J_1 = 7.0$  Hz, 3H); 1.0–1.5 (m, 4H); 2.0 (brs, 1H); 2.20 (m, 1H); 2.30 (s, 3H); 4.25 (d,  $J_2 = 7.8$ , 1H); 4.8–5.2 (m, 2H); 5.3–5.9 (m, 1H); 7.05 (s, 4H). IR: 3400, 1640, 1030, 1000, 915, 820 cm<sup>-1</sup>. MS: 204 ( $M^+$ , 3.7), 187 (9), 162 (56), 121 (7), 91 (100). *erythro*-Form. <sup>1</sup>H NMR: 4.36 (d,  $J_3 = 6.0$  Hz, 1H). The other data of <sup>1</sup>H NMR, IR and MS are the same as above. Anal. Found: C, 82.47; H, 10.06.  $C_{14}H_{20}O$  (mixture products) calcd.: C, 82.30; H, 9.87%.

## 2-Methyl-4-propyl-5-hexen-3-ol (3b entry j)

From isobutyraldehyde: 144 mg. Mixture products: 285 mg, 91%. B.p. 80–83°C/12 mmHg. *threo*-Form. <sup>1</sup>H NMR: 0.9–1.2 (m, 9H); 1.2–1.8 (m, 5H); 1.8–2.5 (m, 2H); 3.05–3.55 (m, 1H); 4.90–5.40 (m, 2H); 5.5–6.0 (m, 1H). IR: 3400, 1640, 1000, 910 cm<sup>-1</sup>. MS: 156 ( $M^+$ , 0.5), 139 (7), 84 (77), 73 (44), 69 (30), 56 (100). *erythro*-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above.

## 3-Propyl-1,5-heptadien-4-ol (**3b** entry k)

From crotonaldehyde: 140 mg. Mixture products: 285 mg, 93%. B.p. 96-100 ° C/15 mmHg. *threo*-Form. <sup>1</sup>H NMR: 0.9 (t, J = 7.0 Hz, 3H); 1.0–1.5 (m, 4H); 1.5–2.5 (m, 5H); 3.7–4.0 (m, 1H); 4.7–5.9 (m, 5H). IR: 3400, 1640, 1020, 970, 910 cm<sup>-1</sup>. MS: 154 ( $M^+$ , 0.14), 138 (100), 136 (5), 95 (30), 81 (47), 71 (85). *erythro*-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above.

## 1-(4-Chlorophenyl)-2-iso-propyl-3-buten-1-ol (3c entry l)

From 4-chlorobenzaldehyde: 281 mg. Mixture products: 420 mg, 94%. B.p.  $138-142^{\circ}$  C/1 mmHg. *threo*-Form. <sup>1</sup>H NMR: 0.85-1.15 (m, 6H); 1.60 (m, 1H); 2.05 (brs, 1H); 2.40 (m, 1H); 4.65 (d,  $J_1 = 7.5$  Hz, 1H); 5.1-6.1 (m, 3H); 7.3 (s, 4H). IR: 3450, 1640, 1090, 1020, 920, 820 cm<sup>-1</sup>. MS: 224 ( $M^+$ , 0.5), 207 (13), 182 (35), 121 (89), 91 (100). *erythro*-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above.

## 1-(4-Methylphenyl)-2-isopropyl-3-buten-1-ol (3c entry m)

From *p*-tolualdehyde: 240 mg. Mixture products: 385 mg, 95%. B.p. 127–130 ° C/1 mmHg. *threo*-Form. <sup>1</sup>H NMR (CDCl<sub>3</sub>): 0.827 (d,  $J_1 = 6.8$  Hz, 6H); 1.46 (m, 1H); 1.95 (brs, 1H); 2.15 (m, 1H); 2.35 (s, 3H); 4.56 (d,  $J_2 = 8.6$  Hz, 1H); 5.15–5.29 (m, 2H); 5.80 (m, 1H); 7.23 (m, 4H). IR: 3400, 1640, 1040, 1000, 910, 810 cm<sup>-1</sup>. MS: 204 ( $M^+$ , 0.1), 203 (0.5), 187 (42), 131 (19), 122 (100), 105 (20). *erythro*-Form. <sup>1</sup>H NMR: 0.90 (d,  $J_1 = 6.8$  Hz, 6H); 2.22 (m, 1H); 4.68 (d,  $J_3 = 8.0$  Hz, 1H); 4.85–4.98 (m, 2H); 5.36 (m, 1H). The other data of <sup>1</sup>H NMR, IR and MS are the same as above.

## 2-Methyl-4-isopropyl-5-hexen-3-ol (3c entry n)

From isobutyraldehyde: 144 mg. Mixture products: 295 mg, 95%. B.p. 78–82°C/12 mmHg. *threo*-Form. <sup>1</sup>H NMR: 0.80–1.0 (m, 12H); 1.2–1.5 (m, 2H); 1.9–2.4 (m, 2H); 3.0–3.3 (m, 1H); 4.9–6.0 (m, 3H). IR: 3400, 1640, 1010, 910 cm<sup>-1</sup>. MS: 156 ( $M^+$ , 0.1), 113 (2), 84 (69), 73 (52), 69 (100), 55 (56), 43 (73), 41 (53). *erythro*-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above. Anal. Found: C, 76.82; H, 13.34. C<sub>10</sub>H<sub>20</sub>O (mixture products) calcd.: C, 76.80; H, 12.90%.

## 3-Isopropyl-1,5-heptadien-4-ol (3c entry o)

From crotonaldehyde: 140 mg. Mixture products: 285 mg, 93%. B.p.  $95-98^{\circ}$  C/15 mmHg. *threo*-Form. <sup>1</sup>H NMR: 0.9–1.6 (m, 7H); 1.7–2.4 (m, 5H); 4.0–4.4 (m, 1H); 5.10–6.20 (m, 3H). IR: 3400, 1640, 1020, 910 cm<sup>-1</sup>. MS: 154 ( $M^{+}$ , 0.2), 153 (2), 138 (100), 136 (5), 95 (18), 81 (44), 71 (41). *erythro*-Form. The same <sup>1</sup>H NMR, IR and MS spectra as above. Anal. Found: C, 77.71; H, 11.73. C<sub>10</sub>H<sub>18</sub>O (mixture products) calcd.: C, 77.87; H, 11.76%.

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#### **References and notes**

- 1 P.A. Bartlett, Tetrahedron, 36 (1980) 3.
- 2 R.W. Hoffmann, Angew. Chem., Int. Ed. Engl., 21 (1982) 555.
- 3 (a) C. Servens and M. Pereyre, J. Organomet. Chem., 35 (1972) C20; (b) Y. Yamamoto, H. Yatagai, Y. Ishihara, N. Maeda and K. Maruyama, Tetrahedron, 40 (1984) 2239.
- 4 (a) Y. Butsugan, H. Ito and S. Araki, Tetrahedron Lett., 28 (1987) 3707; (b) C. Chen, Y.-Z. Huang and Y.C. Shen, Tetrahedron Lett., 29 (1988) 1395.
- 5 The reaction of crotyltin with aldehyde results in  $\alpha$ -adducts as by-products in the presence of BF<sub>3</sub>·Et<sub>2</sub>O and as major products in the presence of AlCl<sub>3</sub>-<sup>i</sup>PrOH. The formation of  $\alpha$ -adducts was due to trans-metallation via  $S_E 2'$  process followed by a rapid reaction with aldehydes. Y. Yamamoto and K. Maruyama, J. Organomet. Chem., 284 (1985) C45.
- 6 M. Kiva, M. Kobayashi and H. Sakurai, Tetrahedron Lett., 28 (1987) 4081.